# THE VISIBLE WAVELENGTH DIFFRACTION FROM TWO-DIMENSIONAL GRATING FABRICATED BY MILD ANODIZATION TECHNIQUE

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Abstract. Two-dimensional diffraction grating was fabricated on aluminum plate on the basis of anodization technique. The reflective grating with interpore ranging 500 nm was prepared using mild anodization. In order to make the porous alumina arrays a single step mild anodization in 0.4 M phosphoric acid with 195 anodization voltage was employed. Interpore distances were seen 500 nm when the mild anodization technique was used. The domain size of hexagonal ordered arrays was observed with over 25  $\mu^2$  area. Optical properties of gratings were investigated using visible light sources thereby the structural details of gratings were determined.

#### **1. Intorduction**

Self organized porous alumina nanostructures fabricated by the anodization of aluminum have attracted considerable attention in electronic, magnetic, photonic, and biosensors devices [1-10]. To facilitate different applications, fabricating of highly ordered porous anodic alumina films with simple process and low cost in a large scale is an essential and urgent. Self ordered porous aluminum oxide (PAO) are obtained by mild anodization (MA) in three regimes with  $H_2SO_4$ ,  $H_2C_2O_4$ , and  $H_3PO_4$  acids at 40 V, 25 V, and 195 V respectively [11-13]. It is found that the formation of a periodic pore structure is sensitive to anodizing conditions. The type of electrolyte (sulfuric acid, phosphoric acid or oxalic acid), applied voltage and anodizing temperature can affect the pore distribution. Though various electrolytes and applied voltages can produce different pore sizes and interpore distances, it has been found that the pore distribution with a high symmetry can be achieved at lower anodizing temperatures [13]. Explanations of the pore formation have been proposed according to the experimental observations of porous structures under various conditions [12, 14].

Diffraction gratings have played a vital role in many fields of science and technology, including spectroscopy, astronomy, beam-sampling for high power lasers, and filtering [15-19]. The most commonly used grating is the metallic reflection grating. One of the main requirements for this kind of grating is the so-called blazing effect, e.g., the concentration of the reflected light in one scattered beam only, at a given wavelength. High-efficiency gratings are desirable for several reasons. A grating with high efficiency is more useful than one with lower efficiency in measuring weak transition lines in optical spectra. A grating with high efficiency may allow the reflectivity and transmissivity specifications for the other

studied. The self-ordering is correlated to the FWHM of the intensity curve against diffracted angle and the smaller FWHM is an indication of the better self-ordering. The degree of self-ordering is directly proportional to intensity of the spots and inversely proportional to the spot size of FFT patterns. Current curve versus time for mild anodization in 0.4 M phosphoric acid at 195 V and 0 °C for 20 hours is shown in Fig. 2.



**Fig. 1.** The visible wavelength diffraction measurements utilized a sample mounted on a rotation stage with an adjustable aperture detector mounted on a rotating arm.



**Fig. 2.** Current curve versus time for mild anodization in 0.1 M phosphoric acid at 195 V and 0 °C for 20 hours.

### 116



Fig. 3. The cross section SEM micrographs of nanopore configurations of sample made in 0.4 M phosphoric acid at the anodization voltage of 195 V. The scale size is  $2 \mu$ .



Fig. 4. 5 μm×5 μm SEM images, FFT and diffraction curve of anodized samples surface in (a) 0.1 M phosphoric acid at 195 V,
(b) 0.4 M phosphoric acid electrolyte at the anodization voltage of 195 V.

In Figure 3 the cross section SEM micrographs of nanopore configurations of sample made in 0.4M phosphoric acid at the anodization voltage of 195 V. The scale size is 2  $\mu$ . SEM micrographs, diffraction curves and FFT images of two samples fabricated in 0.1 M and 0.4 M phosphoric acid at anodization voltage of 195 V are shown in Fig. 3. The FFT pattern is a spotty ring for the sample made in 0.1 M phosphoric acid as seen in Fig. 4 which interprets that the domain size is relatively small and the arrayness of the sample is low. As seen with an increase in phosphoric acid concentration the FFT image changes to a six-fold symmetry

pattern with relatively low spot size which implies an increase in self-ordering. We see a good agreement between optical evaluation and FFT pattern. As shown the diffracted intensity peak reduces to 10 from 12.5 and the FWHM increase to 16 from 14 degree for the samples made with 0.4 M and 0.1 M phosphoric acid electrolyte respectively.

# 4. Conclusion

We have observed self-organization of two-dimensional pore arrays with about 500 nm interpore distances in porous anodic alumina. The experimental results have shown that the electrolyte concentration have a significant effect on the quality of the nanoporous arrays. We see a good agreement between optical evaluation and FFT pattern.

# References

- [1] H. Masuda, M. Satoh // Japan. J. Appl. Phys. 35 (1996) L126.
- [2] J. Liang, H. Chik, A. Yin, J.M. Xu // J. Appl. Phys. 91 (2002) 2544.
- [3] H. Masuda, M. Ohya, H. Asoh, M. Nakao, M. Nohtomi, T. Tamamura // Japan. J. Appl. Phys. 38 (1999) L1403.
- [4] I. Mikulskas, S. Juodkazis, R. Tomaxiunas, J.G. Dumas // Adv. Mater. 13 (2001) 1574.
- [5] R.B. Wehrspohn, J. Schiling // J MRS Bull. 26 (2001) 623.
- [6] H. Masuda, M. Ohya, H. Asoh, K. Nishio // Japan. J. Appl. Phys. 40 (2001) L1217.
- [7] F. Matsumoto, K. Nishio, H. Masuda // Adv. Mater. 16 (2004) 2105.
- [8] K. Nielsch, F. Muller, A.P. Li, U. Gösele // Adv. Mater. 12 (2000) 582.
- [9] K. Nielsch, R. Hertel, R.B. Wehrspohn, J. Barthel, J. Kirschner ,U. Gösele, S.F. Fischer, H. Kronmüller // *IEEE Trans. Magn.* **38** (2002) 2571.
- [10] L. Menon, S. Bandyopadhyay, Y. Liu, H. Zeng, D.J. Sellmyer // J. Nanosci. Nanotech. 1 (2001) 149.
- [11] H. Masuda, K. Fukuda // Science 268 (1995) 1466.
- [12] A. P. Li, F. Muller, A. Birner, K. Nielsch, U. Gosele // J. Appl. Phys. 84 (1998) 6023.
- [13] H. Masuda, K. Yada, A. Osaka // Japan. J. Appl. Phys. 37 (1998) L1340.
- [14] O. Jessensky, F. Muller, U.Gosele // Appl. Phys. Lett. 72 (1998) 1173.
- [15]R. Petit, Electromagnetic theory of gratings (Springer-Verlag, Berlin, 1980).
- [16] M.C. Hutley, *Diffraction gratings* (Academic press, London, 1982).
- [17] D. Maystre, Diffraction gratings (SPIE Milestones series, Gdansk-Jurata, Poland, 1992).
- [18] E.G. Loewen, E. Popov, *Diffraction gratings and applications* (Marcel Dekker, New York, 1997).
- [19] D. Maystre, In: Progress in Optics, ed. by E. Wolf (North-Holland, Amsterdam, 1984), Volume XXI, p. 1.
- [20] J.W. Strutt (Lord Rayleigh) // Philos. Mag. 47 (1874) 193.
- [21] R. Wood // Phil. Mag. Series 6 20 (1910) 770.

### 118